

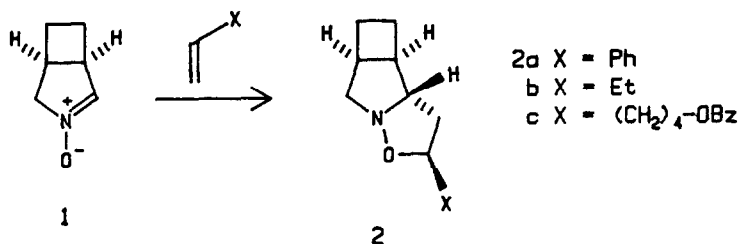
A Highly Stereoselective Synthesis of (E,E)-1,5-Dienes

Joseph J. Tufariello*, Arnold S. Milowsky
Mohammed Al-Nuri and Steven Goldstein
Department of Chemistry
State University of New York at Buffalo
Buffalo, New York 14214

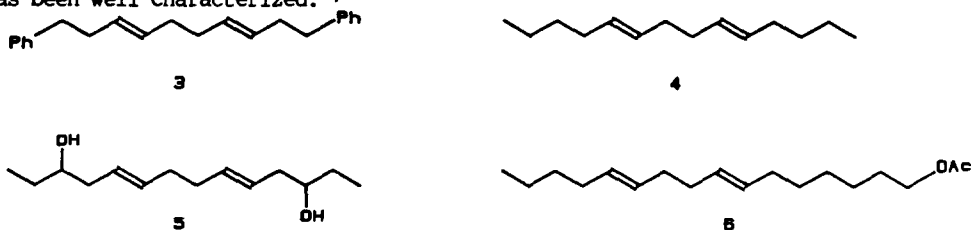
Abstract - Several representative 1,5-dienes were synthesized, with a high degree of regioselectivity, by a series of nitronc cycloadditions and subsequent deamination of the adducts.

Through the use of nitronc cycloaddition chemistry, numerous nitrogenous natural products have been synthesized with excellent stereochemical control;¹ however, very few examples exist which utilize nitrones to generate nitrogen-free products.²⁻⁵ The sesquiterpenes represent a biologically important class of compounds, many of which contain a (E,E)-1,5-diene system. The focus of our recent work has been to synthesize several model compounds containing 1,5-dienes in a highly stereoselective manner by using nitronc cycloaddition chemistry followed by a stereospecific deamination process.

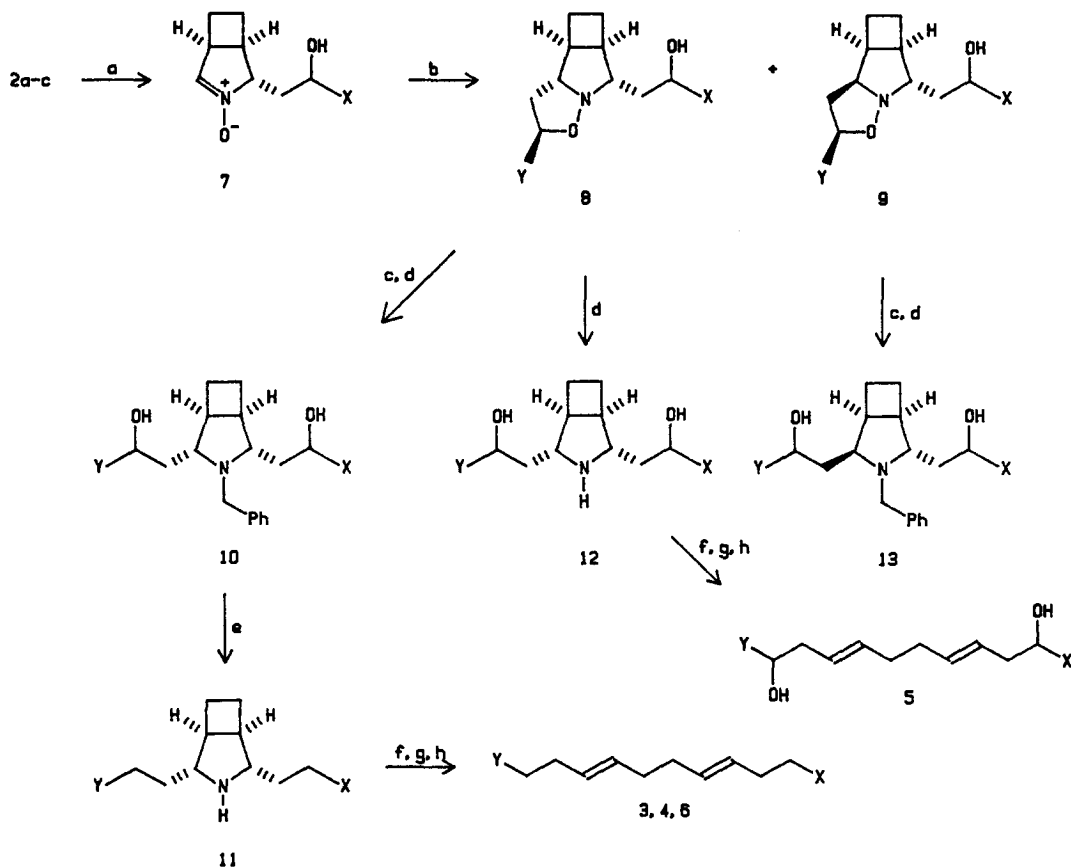
In the preceding communication, we described the synthesis and stereospecific cycloaddition reactions of the bicyclic nitronc **1**, leading to adducts in which the product was formed by cycloaddition from the α -face of the nitronc.⁶



Further manipulation of these adducts allowed us to synthesize the 1,5-dienes, **3-6**, with a high degree of stereoselectivity, where **6** is an isomer of the pink bollworm moth sex pheromone which has been well characterized.^{7,8}



Our general synthetic route is outlined in Scheme I.



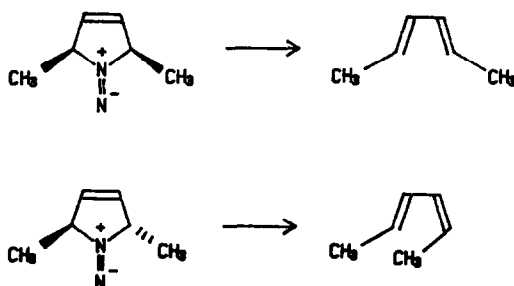
Scheme 1: a. MCPBA; b. H₂C=CH-Y; c. BzBr; d. Zn/HOAc; e. see text; f. HO₃SONO; g. LiAlH₄; h. HgO

In order to further functionalize our system the initial adducts were reconverted into nitrones and permitted to react with a second dipolarophile. The oxidative ring opening reaction of isoxazolidine adducts of this type has been shown to produce the less substituted of the two possible nitrones (ie, nitron 7) with high regioselectivity. Evidence for the formation of this nitron is provided by the presence of the proton signal at δ 6.90–7.00 ppm. The cycloaddition of the nitrones 7a-c with the required dipolarophiles produced separable mixtures of the two stereoisomeric adducts resulting from the addition to the α - or β -face of the nitron. The reaction of 7a with styrene (Y = Ph), produced a 67% yield of 8a and 9a in a 9:1 ratio. The reaction of 7b with 1-butene (Y = Et), produced a 77% yield of 8b and 9b in a 3:1 ratio. Similarly, the reaction of 7c with 1-butene (Y = Et) afforded a 77% yield of 8c and 9c in a 7:3 ratio. The cycloaddition reactions proceeded with good stereoselectivity, providing us with the desired α -face adducts (ie. 8a-c).

Adducts **8a-c** and **9a-c** were independently N-benzylated with benzyl bromide in greater than 90% yield, and the isoxazolidinines were reductively opened with zinc in acetic acid in almost quantitative yield to produce the N-benzyl amino diols **10a-c** and **13a-c**. NMR analysis of the benzylic methylene protons of these compounds confirmed the stereochemical relationship of the side chains. It has been previously shown that a cis relationship between the side chains on a 5-membered ring produces a singlet for the N-benzyl methylene protons, while a trans relationship results in the appearance an AB quartet.^{9,10} Compounds **10a-c** exhibit a singlet attributable to the methylene protons at 3.86, 3.83 and 3.90 ppm respectively. The diols with a trans side chain relationship, **13a-c**, show AB quartets centered about 3.64, 3.68 and 3.70 ppm, respectively, with coupling constants all in the range of 12-14 Hz. Aside from confirming the relative stereochemistry of the side chains, the N-benzyl group also protected the nitrogen during the dehydroxylation sequence. Mesylation of **10b** and **10c** with methanesulfonyl chloride and triethylamine gave the dimesylates (91% from **10b** and 97% from **10c**). An S_N2 displacement by Super Hydride efficiently removed the hydroxyl groups in 72 and 86% respective yield. Removal of the N-benzyl protecting group (and the O-benzyl group for **10c**) was accomplished by hydrogenation in acidic ethanol with a mixture of 10% Pd/C and PdO at 1500 psi, thereby providing **11b** and **11c**. The hydroxyl groups could not be removed from **10a** in the same manner due to the instability of the benzylic mesylate.¹¹ Alternatively, it has been reported that benzylic hydroxyls can be removed by hydrogenation in acidic media using a palladium catalyst.¹² This procedure removed both the hydroxyl groups and the N-benzyl group to give **11a** in 75% yield.

The crucial reaction sequence in our synthesis involved the deamination of **11a-c** in a stereospecific manner to give the all trans 1,5-dienes. It has been demonstrated that diazenes formed from 3-pyrrolines undergo a stereospecific sigmasymmetric fragmentation, producing 1,3-dienes¹³ as shown in Scheme II.

Scheme II



The amino compounds **11a-c** were N-nitrosated, using nitroso sulfuric acid,¹⁴ and converted to the corresponding N-amino compounds by reduction with lithium aluminum hydride. Oxidation of the hydrazines using yellow mercuric oxide¹⁵ provides the all trans-1,5-dienes **3**, **4** and **6** (after acetylation with acetyl chloride and pyridine) as the sole products formed in greater than 90% overall yield for all three cases. The structures and stereochemistry of **3** and **4** were confirmed by an independent synthesis of the corresponding diacetylenic compounds and subsequent reduction

with sodium and ammonia¹⁶ to provide the trans,trans dienes. Partial reduction with Pd on BaSO₄ in the presence of quinoline¹⁷ afforded the cis,cis isomers. Analysis of the ¹³C-NMR data for **3** and **4** showed it to be identical to the ¹³C-NMR data for the dienes obtained by the sodium/ammonia reduction, but drastically different from the partially hydrogenated products obtained from the diacetylene precursors, unambiguously confirming that we had synthesized the (E,E)-dienes.¹⁸ All four isomers of the pheromone, (Z,E)-7,11-hexadecadienyl-1-acetate, have been previously synthesized and fully characterized. Analysis of the ¹³C-NMR data of **6** indicated it corresponded to the (E,E)-isomer.¹⁹

Reductive cleavage of **8b** with zinc and acetic acid afforded the amino diol **12** quantitatively. When **12** was subjected to the nitrosation, reduction, oxidation sequence, the dihydroxy diene **5** was exclusively obtained in 61% yield, demonstrating that functionality can be incorporated into various parts of the diene. We are currently using this method as an integral part of a synthetic approach directed toward several sesquiterpenes.

Acknowledgement:

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18. ¹³C-NMR data; for **3**: 32.6, 34.4, 36.2, 125.7, 128.3, 128.5, 129.7, 130.5, 142.2; for (E,E) structure proof compound: 32.6, 34.4, 36.1, 125.7, 128.2, 128.4, 129.7, 130.4, 142.1; for (Z,Z) structure proof compound: 27.2, 29.2, 36.0, 125.8, 128.1, 128.4, 129.1, 129.9, 142.1; for **4**: 13.9, 22.2, 31.9, 32.3, 32.8, 129.8, 130.7; for (E,E) structure proof compound: 13.9, 22.2, 31.9, 32.3, 32.8, 129.8, 130.7
19. ¹³C-NMR data; for **6**: 13.9, 20.9, 22.1, 25.9, 28.7, 28.7, 29.5, 31.9, 32.2, 32.4, 32.7, 32.7, 64.6, 129.7, 130.0, 130.4, 130.7, 169.5; for 7,11-(E,E)-hexadecadienyl-1-acetate⁸: 13.9, 20.9, 22.2, 25.9, 28.7, 28.7, 29.5, 31.8, 32.3, 32.3, 32.8, 32.8, 64.5, 129.6, 129.8, 130.2, 130.5, 170.7

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